



EXPERIMENTS DIRECTED TOWARD STEREOCONTROLLED SYNTHESIS OF O-LINKED GLYCAN WHICH CONTAINS REPEATING LACTOSAMINE UNIT¹

Zhi-Guang Wang^a, Yukishige Ito^a, Yoshiaki Nakahara^a, and Tomoya Ogawa^{*a,b}

^a The Institute of Physical and Chemical Research (RIKEN), Wako-shi, Saitama, 351-01, Japan

^b Department of Cellular Biochemistry, University of Tokyo, Yayoi, Bunkyo-ku, Tokyo, 113, Japan

Abstract: Repeating lactosamine containing mucin-type oligosaccharide was synthesized as a serine-linked form in a stereocontrolled manner.

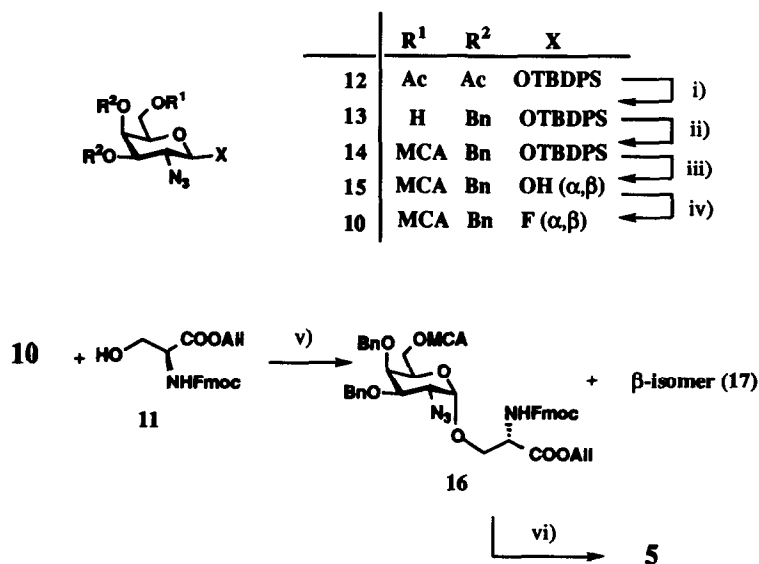
Chemical synthesis of glycoprotein and glycopeptide is a subject of active investigations.² Among structurally diverse glycoprotein oligosaccharides, those containing repeating N-acetylglucosamine (GlcNAc) structures (polylactosamine) attract much attention in connection with various biological events, most notably as onco-differentiation markers.³ In addition to widespread occurrence in asparagine-linked glycoproteins and glycosphingolipids, polylactosamine structures have been revealed to exist in various O-linked (mucin type) oligosaccharides.⁴ As members of this subgroup, oligosaccharides of structure **1** were found from a variety of sources such as human skim milk mucin, equine chorionic gonadotropin and human secretory immunoglobulin A hinge.⁵ A salient character of **1** is that a linearly extended backbone with repeating GlcNAc is attached to GalNAc via 6-OH, forming so called core II structure. As part of our project on the synthesis of mucin-type oligosaccharide with biological significance,⁶ we report here the synthesis of backbone structure of such oligosaccharides in a serine-linked form (i.e. **2**).



1

Our synthetic plan is depicted in **Scheme 1**. The GalNAc residue was designed as an azide **10** so that the α -glycosidic linkage can be constructed with exclusion of participation from C-2 substituent. Since it was anticipated that the stereoselective coupling with serine should be most easily achieved at the stage of monosaccharide,⁷ GlcNAc units were planned to be incorporated onto GalNAc-serine **5**. It is well established in our previous studies on polylactosamine-type glycosphingolipids⁸ that the introduction of GlcNAc units can be achieved under complete stereochemical control in a high yield by taking advantage of strong 1,2-trans directing nature of a NPhth group⁹ and highly efficient activation of anomeric fluoride.¹⁰ Based on such considerations, two GlcNAc units were designed as **6** and **7**.

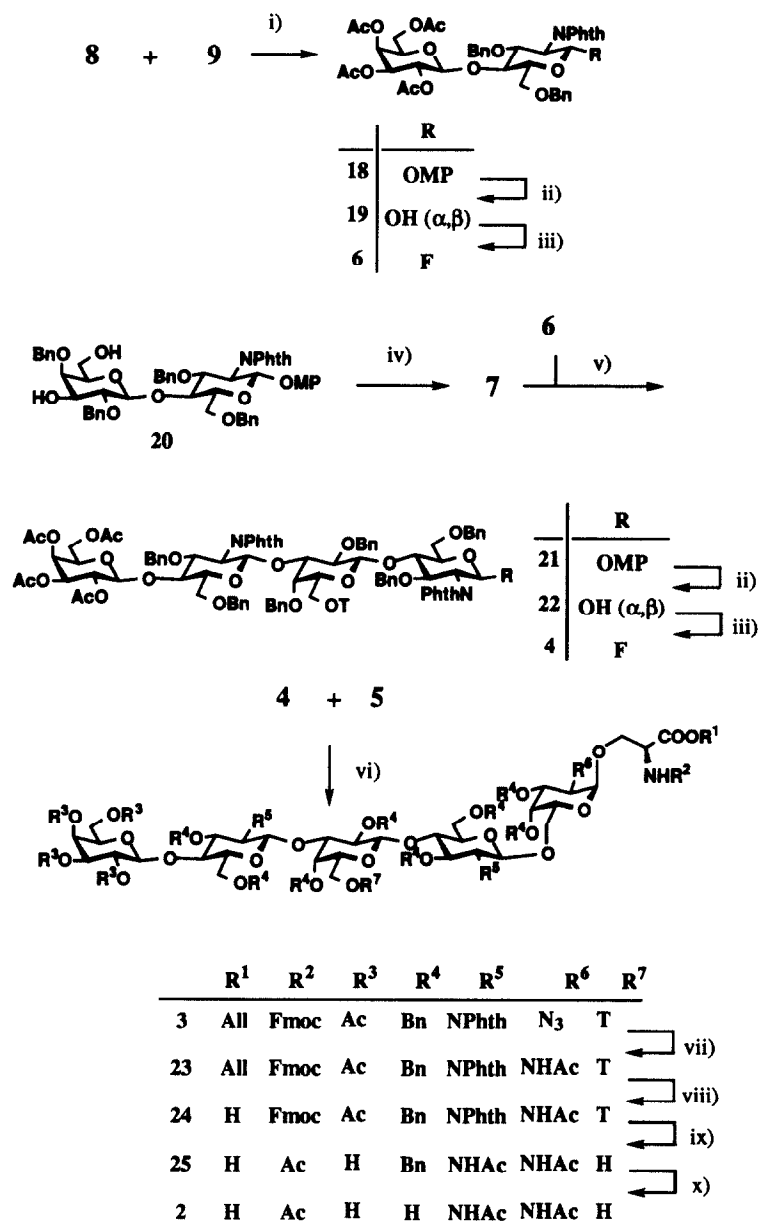
$\text{Cp}_2\text{ZrCl}_2/\text{AgClO}_4^{10a}$ afforded desired α -glycoside **16**¹¹ together with corresponding β -isomer **17**¹¹ in 84 % yield ($\alpha/\beta = 3.33$). Further conversion into **5** was achieved by treatment with thiourea¹⁴.



i) a) NaOMe / MeOH-toluene, r.t.; b) DMTr-Cl / pyridine, r.t.; c) BnBr, NaH / THF, 60°C; d) TsOH / MeOH-CHCl₃; 80% overall. ii) (ClCH₂CO)₂O, DMAP / CH₂Cl₂-Et₃N; 81%. iii) *n*-Bu₄NF / THF-AcOH, r.t.; 97%. iv) DAST / (ClCH₂)₂; 90%. v) AgClO₄- Cp₂ZrCl₂ (2:1) / (ClCH₂)₂, -23°C; 84%. vi) thiourea / DMF, 90°C; 58%.

Scheme 2

Construction of tetrasaccharide donor **4**¹¹ and coupling with **5** were performed as follows (**Scheme 3**). Glycosylation of the bromide **8** with **9** afforded **18**¹¹ in 69% yield. Subsequent removal of 4-methoxyphenyl group¹⁵ gave hemiacetal **19**¹¹ and fluorination¹⁶ furnished donor **6**¹¹. Selective acylation of **20**¹⁷ was cleanly achieved with toluoyl chloride at 0°C. Resulting **7**¹¹ was glycosylated with **6** by using Cp₂HfCl₂-AgOTf^{10b} as a promoter to produce **21**¹¹. After conversion into **4**, coupling with **5** was performed again under Suzuki's conditions.^{10b} The reaction proceeded smoothly to give the coupling product **3**¹¹ in good yield, while SnCl₂-AgOTf¹⁸ promoted reaction required longer reaction time and gave lower yield (ca. 50%).



i) AgOTf / (CH₂Cl)₂, -15°C-r.t.; 69%. ii) CAN / toluene-MeCN-H₂O; 85-91%. iii) DAST / (CH₂Cl)₂, 0°C; 85-91%. iv) toluoyl chloride / (CH₂Cl)₂-pyridine, 0°C; 93%. v) Cp₂HfCl₂-AgOTf (1:2) / CH₂Cl₂, -23°C-0°C; 87%. vi) Cp₂HfCl₂-AgOTf / (CH₂Cl)₂, -25°C-r.t.; 79% or AgOTf-SnCl₂ / (CH₂Cl)₂, r.t.; 50%. vii) MeCOSH / pyridine, r.t.; 89%. viii) Pd(PPh₃)₄, MeNHPh / THF; 93%. ix) NH₂NH₂-H₂O / EtOH, 80°C then Ac₂O / aq. dioxane, r.t.; 38%. x) H₂, Pd(OH)₂ / MeOH-H₂O-AcOH (40:10:1); 87%.

Scheme 3

Further transformation into **2** was conducted as follows. One-step conversion of azide **3** to acetamide **23**¹¹ was realized by treatment with AcSH/Pyridine (2:1 volume)¹⁹. Subsequent deallylation into carboxylic acid **24** was accomplished with Pd(PPh₃)₄ and MeNHPh.²⁰ Exposure of **24** to hydrazine/ethanol (1:2) at 80°C for 3 days followed by N-selective acetylation with 40 equiv. of acetic anhydride in aq. dioxane afforded **25**. Hydrogenolysis with 20% Pd(OH)₂/C proceeded smoothly to produce **2**¹¹ as a white foam after lyophilization. Spectrum data (NMR, MS) are well in agreement with the structure.

As is described above, we have established a reasonably practical synthetic route to O-linked glycan chain which contains repeated LacNAc units. This strategy, after slight modification, should be applicable to glycan chain with properly functionalized amino acid residue, which can be incorporated into peptide sequences.

Acknowledgments. Z.-G. W. would like to thank Meiji Milk Products Co. Ltd. for postdoctoral fellowship. A part of this work was financially supported by a Grant-in Aid for Scientific Research on Priority Areas No. 06240105 from the Ministry of Education, Science and Culture, and also by the Special Coordination Funds of the Science and Technology Agency of the Japanese Government. We thank Dr. J. Uzawa and Mrs. T. Chijimatsu for NMR, Mr. Y. Esumi for FAB-MS, Ms. M. Yoshida and her staff for elemental analyses, and Ms. A. Takahashi for technical assistance.

References and Notes

1. Part 100 in the series of "Synthetic Studies on Cell-surface Glycans". For Part 99 see; Nunomura, S.; Matsuzaki, Y.; Ito, Y.; Ogawa, T. *Pure Appl. Chem.* **1994**, *66*, 2123.
2. For reviews see: (a) Kunz, H. *Angew. Chem. Int. Ed. Engl.* **1987**, *26*, 294. (b) Paulsen, H.; Merz, G.; Peters, S.; Weichert, U. *Liebigs Ann. Chem.* **1990**, 1165.
3. Fukuda, M. *Biochem. Biophys. Acta* **1984**, *780*, 119.
4. Jontoft, N. *Trends Biochem. Sci.* **1990**, *15*, 291; Carraway, K. L.; Hull, S. R. *Glycobiology* **1991**, *1*, 347; Fukuda, M. *ibid.* **1991**, *1*, 131.
5. Hanisch, F.G.; Uhlenbruck, G.; Peter-Katalinic, J.; Egge, H.; Dabrowski, J.; Dabrowski, U. *J. Biol. Chem.* **1989**, *264*, 872; Kokke, C. H.; Roosenboom, M. J. H.; Thomas-Oates, J. E.; Kamerling, J. P.; Vliegthart, J. F. G. *Glycoconjugate J.*, **1994**, *11*, 35; Pierce-Cretel, A.; Decottignies, J. P.; Wieruski, J.M.; Strecker, G.; Montreuil, J.; Spik, G. *Eur. J. Biochem.* **1989**, *182*, 457.
6. For the previous paper, see; Nakahara, Y.; Iijima, H.; Ogawa, T. *Tetrahedron Lett.* **1994**, *20*, 3321.
7. Ferrai, B.; Pavia, A. A. *Carbohydr. Res.* **1980**, *79*, C1.
8. Matsuzaki, Y.; Ito, Y.; Nakahara, Y.; Ogawa, T. *Tetrahedron Lett.* **1993**, *34*, 1066.
9. Lemieux, R. U.; Takeda, T.; Chung, B. Y. *ACS Symp. Ser.* **1977**, *39*, 90.
10. a) Matsumoto, T. Maeta, H.; Suzuki, K.; Tsuchihashi, G. *Tetrahedron Lett.* **1988**, *29*, 3567. b) Suzuki, K.; Maeta, H.; Matsumoto, T. *ibid.* **1989**, *30*, 4853.
11. Selected physical data for key compounds are given below. Values of $[\alpha]_D$ and ¹H- and ¹³C-NMR were recorded at 25±3°C for solutions in CHCl₃ or CDCl₃ respectively, unless otherwise indicated. **2**: R_f 0.33 in 1:1:1:1 nBuOH-EtOH-H₂O-HOAc; $[\alpha]_D$ +19° (c 0.1, H₂O); δ_H (D₂O, 70°C) 4.83(d, J 3.7 Hz, 1H, H-1a), 4.73 and 4.59(2d, J 7.7 Hz, 7.3 Hz, 2H, H-1b, and H-1d), 4.46 and 4.45(2d, J 7.6 Hz and 8.1 Hz, 2H, H-1c and H-1e), 4.15 (dd, J 4.0 and 11 Hz, 1H, H-2a), 4.13(d, J 4.0 Hz, 1H, H-4a), 2.04, 2.03, 2.02, 2.01(4s, 12H, 4Ac); Positive FAB-MS 1081(M), 1104(M+Na); Negative FAB-MS 1080(M-

- 1). **3**: Rf 0.36 in 2:1 toluene-EtOAc; $[\alpha]_D^{+34^\circ}$ (c 0.4); δ_H 5.82(m, 2H, =CH, NH), 5.36(d, J 8.6 Hz, 2H, H-1b and 1c), 5.28(dd, J 1.4 and 15.2 Hz, 1H, =CH₂), 5.26(dd, J 2.9 Hz, H-4e), 5.20(dd, J 1.5 and 10 Hz, 1H, =CH₂), 5.14 (dd, J 7.9 and 10.3 Hz, H-2e), 2.35(s, 3H, CH₃-C₆H₄), 2.062, 2.012, 1.989, and 1.977(4s, 12H, 4 Ac); δ_C 102.89, 100.31, 99.84, 99.81, 99.25. **4**: Rf 0.46 in 1:1 hexane-EtOAc; $[\alpha]_D^{+17^\circ}$ (c 0.4); δ_H 5.70(dd, J 7.9 and 54 Hz, 1H, H-1a), 5.40(d, J 8.2 Hz, 1H, H-1c), 5.27(d, J 3.3 Hz, 1H, H-4d), 5.18(dd, J 10.6 and 7.9 Hz, 1H, H-2d), 2.065, 2.021, 1.997, and 1.978(4s, 12H, 4Ac). **7**: Rf 0.33 in 1:1 hexane-EtOAc; $[\alpha]_D^{+50.5^\circ}$ (c 0.5) δ_H 5.84(dd, J 54 and 7.3 Hz, 1H, H-1a), 5.27(d, J 2.7 Hz, 1H, H-4b), 5.15(dd, J 7.9 and 10.6 Hz, 1H, H-2b), 4.45(d, J 7.9 Hz, 1H, H-1b), 2.078, 2.063, 2.043, and 2.011(4s, 12H, 4 Ac). **10 α** : Rf 0.74 in 2:1 Hexane-EtOAc; $[\alpha]_D^{+48.7^\circ}$ (c 0.67); δ_H 5.62(dd, J 2.4 and 53 Hz, 1H, H-1). **10 β** : Rf 0.65 in 2:1 hexane-EtOAc; $[\alpha]_D^{-20.1^\circ}$ (c 1); δ_H 4.96(dd, J 7.6 and 53.5 Hz, 1H, H-1), 3.97(s, 2H, COCH₂Cl). **16** Rf 0.66 in 3:1 toluene-EtOAc; $[\alpha]_D^{+56^\circ}$ (c 0.6); δ_H 5.91(m, 1H, CH=), 5.33(dd, J 17.2 and 1.3 Hz, 1H, =CH₂), 5.25(dd, J 10.6 and 1 Hz, =CH₂), 4.80(d, J 2 Hz, 1H, H-1), 3.95(s, 2H, COCH₂Cl); δ_C 99.64 (C-1). **17**: Rf 0.29 in 3:1 toluene-EtOAc; $[\alpha]_D^{-4.8^\circ}$ (c 0.4); δ_H 5.91(m, 1H, CH=), 5.33(dd, J 1.3 and 17.2 Hz, 1H, =CH₂), 5.22(dd, J 1.3 and 10.6 Hz, =CH₂), 4.12(d, J 7.3 Hz, 1H, H-1), 3.90(s, 2H, COCH₂Cl); δ_C 102.6 (C-1). **18**: Rf 0.40 in 2:1 hexane-EtOAc; $[\alpha]_D^{+60^\circ}$ (c 0.35); δ_H 5.62(d, J 8.3 and 10.5 Hz, H-2b), 5.30(d, J 2.9 Hz, 1H, H-4b), 5.18(dd, J 8 and 10.5 Hz, H-2b), 4.50(d, J 9.2 Hz, H-1b), 3.70(s, 3H, OCH₃), 2.056, 2.039, 2.036, and 2.029(4s, 12H, 4Ac) δ_C 100.04(C-1b), 97.5(C-1a). **21**: Rf 0.25 in hexane-EtOAc; $[\alpha]_D^{+30.4^\circ}$ (c 2); δ_H 5.44(d, J 8.3 Hz, 1H, H-1a), 5.22(d, J 2 Hz, 1H, H-4d), 5.38(d, J 7.9 Hz, 1H, H-1c), 3.65(s, 3H, OCH₃), 2.39(s, 3H, CH₃-C₆H₄), 2.06, 2.02, 1.99, and 1.97(4s, 12H, 4Ac). **23**: Rf 0.81 in 9:1 CHCl₃-CH₃OH; $[\alpha]_D^{+23.6^\circ}$ (c 0.5); δ_H 5.80(m, 1H, CH=), 5.62(d, J 10 Hz, 1H, NH), 5.37(d, J 8.6 Hz, 2H, H-1b and H-1d), 2.35(s, 3H, CH₃-C₆H₄), 2.066, 2.013, 1.992, 1.979, and 1.824(5s, 15H, 5Ac). **25**: Rf 0.2 in 3:1 CH₃Cl-CH₃OH; $[\alpha]_D^{+28^\circ}$ (c 0.2, CH₃OH); δ_H (CD₃OD) 7.13-7.40(m, aromatic-H), 5.134(d, J 10.7 Hz, 1H, OCH₂Ph), 5.040(d, J 11.7 Hz, 1H, OCH₂Ph), 5.000(d, J 10.2 Hz, 1H, OCH₂Ph), 1.990, 1.984, 1.939, and 1.623(4s, 12H, 4 Ac); Positive FAB-MS 1801(M) 1824(M+Na).
12. Nakahara, Y.; Iijima, H.; Shibayama, S.; Ogawa, T. *Tetrahedron Lett.* **1990**, *34*, 6897; Lemieux, R. U.; Ratcliffe, R. M. *Can. J. Chem.* **1979**, *57*, 1244.
13. de la Torre, B. G.; Torre, J. L.; Bardaji, E.; Clapés, P.; Xaus, N.; Jorba, X.; Clavet, S.; Albericio, F.; Valentia, G. *J. Chem. Soc. Chem. Commun.* **1990**, 965.
14. Naruto, M.; Ohno, K.; Naruse, N.; Takeuchi, H. *Tetrahedron Lett.* **1979**, 251.
15. Fukuyama, T.; Laird A.A.; Hotchkiss, L.M., *Tetrahedron Lett.* **1985**, *26*, 6291.
16. Rosenbrook, Jr., W.; Riley, D. A.; Larty, P. A. *Tetrahedron Lett.* **1985**, *26*, 3; Posner, G. H.; Haines, S. R. *ibid.* **1985**, *26*, 5.
17. Nakano, T.; Ito, Y.; Ogawa, T.; *Carbohydr. Res.* **1993**, *243*, 43.
18. Mukaiyama, T.; Murai, Y.; Shoda, S. *Chem. Lett.* **1981**, 431.
19. Rosen, T.; Lico, I. M.; Chu, D. T. W. *J. Org. Chem.* **1988**, *53*, 1580.
20. Ciommer, M.; Kunz, H. *Synlett.* **1991**, 593.

(Received in Japan 15 October 1994; accepted 31 October 1994)